

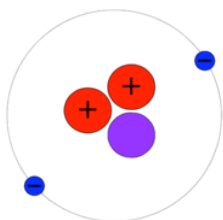
# 8th Grade Distance: Distance Learning Activities

April 4 - April 10 Lesson Plan for Grade 8 Science

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## MONDAY - Sub-Atomic Particles

Draw an arrow labeling the three different parts. Fill in the chart with the proper information.



Particle	Location	Charge	Mass
Proton			
Neutron			
Electron			

## TUESDAY - Vocabulary Review for Chemistry (Periodic Table)

1 H Hydrogen 1.008																	2 He Helium 4.003	
3 Li Lithium 6.94	4 Be Beryllium 9.012											5 B Boron 10.81	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180	
11 Na Sodium 22.990	12 Mg Magnesium 24.305											13 Al Aluminum 26.982	14 Si Silicon 28.085	15 P Phosphorus 30.974	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ar Argon 39.948	
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.867	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.630	33 As Arsenic 74.922	34 Se Selenium 78.97	35 Br Bromine 79.904	36 Kr Krypton 83.798	
37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.95	43 Tc Technetium [97]	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.906	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.414	49 In Indium 114.818	50 Sn Tin 118.710	51 Sb Antimony 121.760	52 Te Tellurium 127.60	53 I Iodine 126.904	54 Xe Xenon 131.293	
55 Cs Cesium 132.905	56 Ba Barium 137.327	* 57 - 70	71 Lu Lutetium 174.967	72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	78 Ir Iridium 192.217	79 Pt Platinum 195.084	80 Au Gold 196.967	81 Hg Mercury 200.592	82 Tl Thallium 204.38	83 Pb Lead 207.2	84 Bi Bismuth 208.980	85 Po Polonium [209]	86 At Astatine [210]	87 Rn Radon [222]
87 Fr Francium [223]	88 Ra Radium [226]	** 89 - 102	103 Lr Lawrencium [262]	104 Rf Rutherfordium [267]	105 Db Dubnium [270]	106 Sg Seaborgium [269]	107 Bh Bohrium [270]	108 Hs Hassium [270]	109 Mt Meitnerium [278]	110 Ds Darmstadtium [281]	111 Rg Roentgenium [281]	112 Cn Copernicium [285]	113 Nh Nihonium [286]	114 Fl Flerovium [289]	115 Mc Moscovium [289]	116 Lv Livermorium [293]	117 Ts Tennessine [293]	118 Og Oganesson [294]
*Lanthanide series		57 La Lanthanum 138.905	58 Ce Cerium 140.116	59 Pr Praseodymium 140.908	60 Nd Neodymium 144.242	61 Pm Promethium [145]	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.500	67 Ho Holmium 164.930	68 Er Erbium 167.259	69 Tm Thulium 168.934	70 Yb Ytterbium 173.045			
**Actinide series		89 Ac Actinium [227]	90 Th Thorium 232.038	91 Pa Protactinium 231.036	92 U Uranium 238.029	93 Np Neptunium [237]	94 Pu Plutonium [244]	95 Am Americium [243]	96 Cm Curium [247]	97 Bk Berkelium [247]	98 Cf Californium [251]	99 Es Einsteinium [252]	100 Fm Fermium [257]	101 Md Mendelevium [258]	102 No Nobelium [259]			

## TUESDAY - Vocabulary Review for Chemistry (Periodic Table) continued.

1. Color / Highlight the three major groups of the Periodic Table above. (Metals....etc)
2. List three properties of the metals and the non-metals
  - a. Metals:
  - b. Non-metals:
  - c. What makes the metalloids unique?
3. What is a group? What is a Period? How many groups and periods are there?

## WEDNESDAY - Question analysis

Answer the following question. Either post to GC or save in your science binder.

1. What do all of the numbers on the element square represent?
2. How do I find the number of neutrons in an atom from the information I have in the square?

## THURSDAY - Making Models

1. Using items from around your house or yard, make a physical model of an atom of Potassium (element #19). Make sure to separate the different electron shells with the proper number of electrons attached in each shell.
2. Explain to a parent or to a sibling what makes up an atom, (the three parts) and their locations.
  - a. Have parents initial here stating that you communicated this. \_\_\_\_\_
3. Either post a picture in the GC or draw your model and save it in your binder.

## FRIDAY - Game Day!!

*The day you all have been waiting for...*

1. Live Kahoot! At 10:00 AM.
2. I will post the game code in the google classroom and send it out through Remind.
3. **You may only play one time** and must include your name as your game name. We are recording who plays each week based on your names.

## STATE STANDARD

Strand 5, Concept 1, PO 6

~ Identify different kinds of matter based on their physical properties.